

Strontium Titanate Perovskite Based Electrode Materials for Solid Oxide Membrane Reactors

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Oxygen is a major component of many catalytic reactions, including, fuel cells, energy-related applications, emission control, and chemical synthesis. Control of selectivity for any chemical reaction is a challenge, and oxygen is known to have a major impact of selectivity of many oxygen containing reactions. Besides the amount of oxygen, oxygen species available also affect the selectivity of the reaction. Solid Oxide Membrane Reactors consisting of an oxide ion conducting electrolyte and two electrodes (anode and cathode) can help improve the selectivity for oxygen containing catalytic reactions, by controlling amount of oxygen available for the reaction. Moreover, tailoring the properties of electrode catalysts, oxygen species on the surface and the selectivity can be controlled.

Strontium titanate type perovskites were tested as electrode materials for these reactors and they were tested for oxidative dehydrogenation of ethane to ethylene and CO₂ & H₂O reduction reactions [1, 2, 3]. The effect of lanthanum doping and chlorine incorporation on these catalysts was investigated. The characteristics of the catalysts were investigated using ambient and in-situ X-ray diffraction (XRD), X-ray photoelectron spectroscopy (XPS), diffuse reflectance infrared spectroscopy (DRIFTS), temperature programmed oxidation (TPO) and electronic conductivity measurements. Besides Electro-ODH experiments to test the activities, impedance measurements were also taken using to better understand the cell characteristics and resistances. Oxygen flux measurements were also performed.

References:

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